

# LEDs and Lasers

Turn off all electronic devices

## Observations about LEDs and Lasers

- LEDs and Lasers often have pure colors
- LEDs can operate for years without failing
- Lasers produce narrow beams of intense light
- Lasers are dangerous to eyes
- Reflected laser light has a funny speckled look

## 6 Questions about LEDs and Lasers

1. Why can't electrons move through insulators?
2. How does charge move in a semiconductor?
3. Why does a diode carry current only one way?
4. How does an LED produce its light?
5. How does laser light differ from regular light?
6. How does a laser produce coherent light?

## Question 1

Q: Why can't electrons move through insulators?  
 A: Electrons can't easily change levels in insulators.

Electrons obey the rules of quantum physics

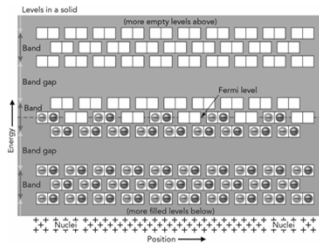
- In matter, electrons exist as quantum standing waves
  - three-dimensional patterns of nodes and antinodes
  - each wave "cycles" in place—it does not change with time
- In solids, those standing waves are called levels

To move, electrons must be able to switch levels

In an insulator, electrons can't easily change levels

## Electrons in Solids

Electrons settle into a solid's lowest-energy levels  
 Levels clump into energy bands, leaving gaps  
 Fermi level: between last filled and first unfilled

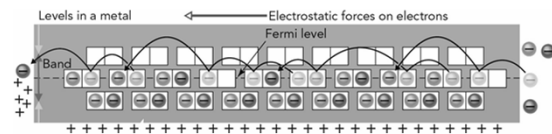


## Metals

In a metal,

- the Fermi level has empty levels just above it in energy
- electrons near the Fermi level can change levels easily
- electrons can move in response to electric fields

The electrons are like patrons in a partly filled theatre  
 Current can flow through a metal



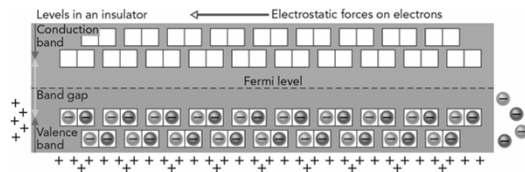
## Insulators

In an insulator,

- the Fermi level has no empty levels nearby
- electrons can't move in response to electric fields

The electrons are like patrons in a completely filled theatre

Current can't flow through an insulator



## Question 2

Q: How does charge move in a semiconductor?

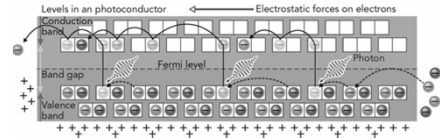
A: A semiconductor is an insulator with a small gap

Pure semiconductors are "poor insulators"

- empty conduction band is just above full valence band

The electrons are like patrons in a theater with a full ground floor but a low empty balcony

- Light or heat can allow current in pure semiconductor



## Doped Semiconductors

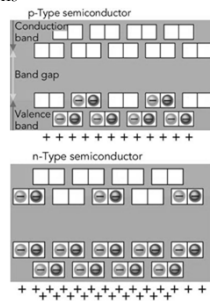
Dopant atoms can add or remove electrons

A p-type semiconductor

- has fewer electrons than normal
- has some empty valence levels

An n-type semiconductor

- has more electrons than normal
- has some filled conduction levels



## Question 3

Q: Why does a diode carry current only one way?

A: Its p-n junction is asymmetric and special

When p-type and n-type semiconductors touch

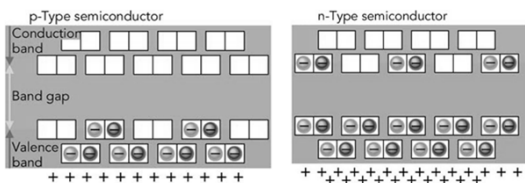
- they have different affinities for electrons
- electrons migrate from the n-type to p-type
- leaving a depletion region with no mobile charges

That junction can conduct current only one way

## pn-Junction (before contact)

Before p-type semiconductor meets n-type,

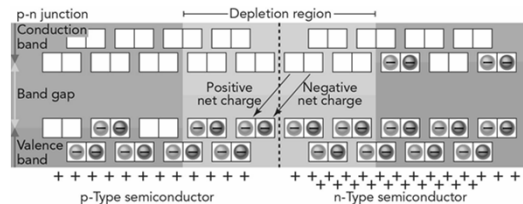
- each material can conduct electricity
- each material is electrically neutral throughout



## pn-Junction (after contact)

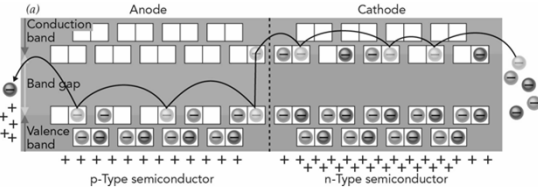
After p-type semiconductor meets n-type,

- electrons migrate from the n-type to the p-type
- an insulating depletion region appears at junction
- depletion region develops an electric field



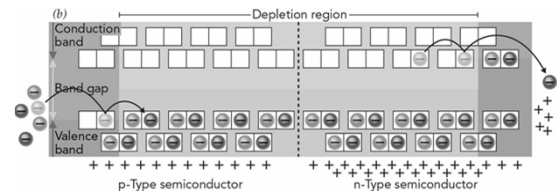
### Forward in the Diode

- When electrons enter n-type and exit p-type,
- depletion region's electric field is cancelled away
  - depletion region shrinks
  - diode can conduct current



### Reverse in the Diode

- When electrons enter p-type and exit n-type,
- depletion region's electric field is reinforced
  - depletion region grows
  - diode cannot conduct current



### Clicker Question

When electrons cross a diode's pn-junction from n-side to p-side, they are in the conduction band. They leave the p-side in the valence band. Shortly after crossing the junction, each electron

- A. releases energy by dropping from the conduction band to the valence band.
- B. absorbs energy by rising from the conduction band to the valence band.

### Question 4

Q: How does an LED produce its light?  
 A: Electrons emit light while changing bands

LEDs are Light-Emitting Diodes

- LED is a diode and has a pn-junction
- Electrons cross junction in the conduction band
- Electrons dropping into the valence band emit light
  - Electron briefly orbits the empty valence level
  - Electron drops into valence level via a radiative transition

The larger the band gap, the bluer the light

### Question 5

Q: How does laser light differ from regular light?  
 A: Laser light is a single electromagnetic wave.

Most light sources produce photons randomly

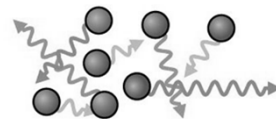
- Each photon usually has its own wave

Laser light involves duplicate photons

- Laser amplification duplicates an initial photon
- Each photon becomes part of a single giant wave

### Spontaneous Emission

Excited atoms normally emit light spontaneously  
 These photons are uncorrelated and independent  
 Each photons has its own wave mode  
 These independent waves are incoherent light



Incoherent radiation from excited atoms

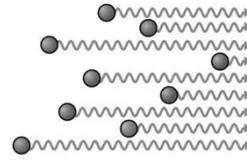
### Clicker Question

If you split the beam from a flashlight into two beams and overlap those beams on a white screen, can you see interference effects?

- A. Yes, because beams are parts of one light wave
- B. Yes, because beams contain many waves
- C. No, because too many independent light waves
- D. No, because beams don't contain waves

### Stimulated Emission

Excited atoms can be stimulated into duplicating passing light  
 These photons are correlated and identical  
 The photons all have the same wave mode  
 This single, giant wave is coherent light



Coherent radiation from excited atoms

### Clicker Question

If you split the beam from a laser pointer into two beams and overlap those beams on a white screen, can you see interference effects?

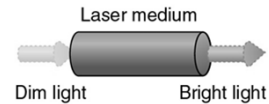
- A. Yes, because beams are parts of one light wave
- B. Yes, because beams contain many waves.
- C. No, because too many independent light waves
- D. No, because beams don't contain waves

### Question 6

Q: How does a laser produce coherent light?  
 A: A laser amplifier duplicates an initial photon.

Excited atom-like systems can act as laser medium

- Duplicate photons they're capable of emitting
- Duplication is perfect: the photons are true clones



### Clicker Question

If you place mirrors around a laser medium,

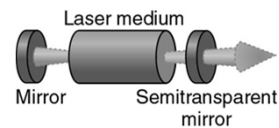
- A. nothing will happen because the mirrors will prevent light from reaching the laser medium.
- B. a photon emitted spontaneously by the laser medium will be duplicated endlessly.

### Laser Oscillation

A laser medium can amplify its own light

- A laser medium in a resonator acts as an oscillator
- It duplicates its one of its own spontaneous photons
- Duplicated photons leak from semitransparent mirror

The photons from this oscillator are identical



## **Summary about Lasers and LEDs**

Lasers produce coherent light by amplification  
Coherent light contains many identical photons  
Laser amplifiers and oscillators are common  
LEDs are incoherent, light-emitting diodes